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Section 5.3 – Compounds and Elements

1) Define Compounds.

- 2) Are compounds pure substances?
- 3) How many types of atoms are in an element?
- 4) What is the smallest unit of a compound called?
- 5) What is the difference between a molecule and an atom?
- 6) What is the difference between an element and a compound?
- 7) Give two examples of an element and two examples of a compound?
- 8) How did ancient Greek Philosophers think about matter?
- 9) Where did most scholars think that wisdom came from?
- 10) Who were alchemists?

11) List 4 characteristics about them.

12) When did the current view of the elements begin to develop?

13) Who made a very important contribution with his use of a scientific method to investigate the physical world?

14) What did Bacon's book argue about the nature of science?

15) What was Robert Boyle's contribution?

16) Who defined elements as pure substances that cannot be decomposed (broken down) into simpler substances by means of a chemical change?

17) How many pure substances did he define as elements?

18) What was Lavoisier's most successful technique as an experimenter?

Adapted from Science Power 9 - McGraw-Hill Ryerson